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#### **Key indicators**

Single-crystal X-ray study T = 190 KMean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ Å}$  R factor = 0.027 wR factor = 0.063 Data-to-parameter ratio = 9.8

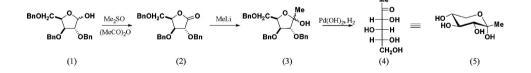
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The crystalline form of 1-deoxy-D-sorbose,  $C_6H_{12}O_5$ , is shown to be 1-deoxy- $\alpha$ -D-sorbopyranose. This is the first reported crystal structure of a 1-deoxyketose. The absolute configuration was determined by the use of D-xylose as the starting material. The crystal structure has a three-dimensional hydrogen-bonded network.

1-Deoxy-*a*-D-sorbopyranose

## Comment

Although the driving force for the large-scale production of rare sugars by biotechnological (Izumori, 2002; Granström et al., 2004) and chemical (Beadle et al., 1992) methods is driven by the demand for alternative foodstuffs (Skytte, 2002), rare monosaccharides such as D-psicose (Takata et al., 2005; Matsuo et al., 2006) and D-allose (Sui et al., 2005; Hossain et al., 2006) have significant chemotherapeutic properties. As well as being useful for their potential biological properties, the 1deoxyketoses are likely to provide a new set of building blocks for the synthesis of a wide variety of complex biomolecules. However, the properties of 1-deoxyketoses have been little studied to date; there are no reports of the crystal structure of any of the four diastereomers. As part of our work to extend the range of simple monosaccharide derivatives, 1-deoxy-Dsorbose, (4), was synthesized. Although the compound has been prepared previously (James & Angyal, 1972; Dills & Meyer, 1976), a solution of the compound contains a mixture of equilibrating structures (Angyal et al., 1976). 1-Deoxy-Dsorbose was readily crystallized and this paper firmly establishes that it exists in the crystalline state as the  $\alpha$ -anomer of the pyranose ring form, (5), in a chair conformation.

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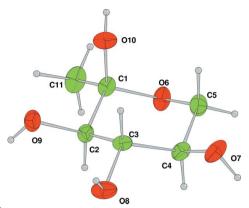


In summary, 1-deoxy-D-sorbose, (4), exists in the crystalline state as 1-deoxy- $\alpha$ -D-sorbopyranose, (5). The absolute configuration was determined by the use of D-xylose as the starting material. A D-sugar is defined by the absolute stereochemistry at C-5 (relative to D-glyceraldehyde); see http://www.chem.qmw.ac.uk/iupac/2carb/ for an explanation of carbohydrate nomenclature (IUPAC–IUBMB, 1996). The present X-ray crystal structure determined the stereochemistry at the anomeric position (C1) as being  $\alpha$ , with the hydroxyl group in the axial position.

The crystal structure of (5) has a three-dimensional hydrogen-bonded network, with each molecule interacting

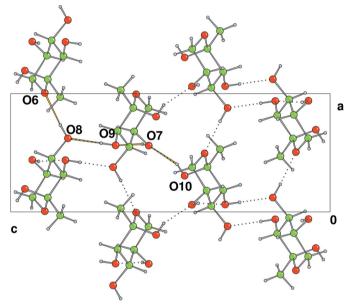
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## Figure 1

The molecular structure of the title compound, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as spheres of arbitrary radii.



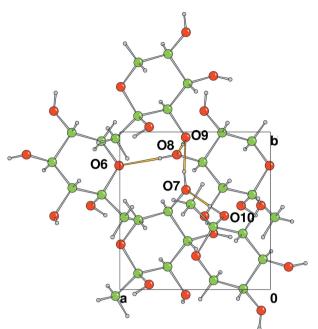
#### Figure 2

The crystal structure of (5), projected along the b axis, showing the threedimensional hydrogen-bonding network (dotted lines). The hydrogenbond chain involving atoms O6, O7, O8, O9 and O10 is highlighted in orange.

with six neighbours (Fig. 2). The hydrogen bonds themselves form a discrete continuous chain:  $O10 \cdots O7$ ,  $O7 \cdots O9$ ,  $O9 \cdots O8$  and  $O8 \cdots O6$ , with O10 at the head of the chain as a donor and O6 at the tail as an acceptor (Fig. 3).

## **Experimental**

For the synthesis of 1-deoxy-D-sorbose, the tribenzylated derivative of D-xylose, (1) (Barker & Fletcher, 1961; Postema *et al.*, 2000), was oxidized to the lactone, (2), with acetic anhydride and dimethyl sulfoxide (Calzada *et al.*, 1995). Addition of methyl lithium to the protected lactone, (2), afforded the lactol, (3). Subsequent hydrogenation yielded 1-deoxy-D-sorbose, (4) (Jones *et al.*, in preparation). The title compound, (5), was recrystallized from a mixture of ethyl acetate and methanol (3:1) to give colourless crystals (m.p. 425–427 K).  $[\alpha]_D^{20}$  50.2 (*c* 1.0 in H<sub>2</sub>O).



#### Figure 3

A projection of the crystal structure along the *c* axis, showing the five molecules linked by the discrete hydrogen-bond chain, in which the  $H \cdots O$  hydrogen bonds are shown in orange.

Z = 4

 $D_x = 1.494 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

Needle, colourless

 $0.60 \times 0.20 \times 0.20$  mm

1613 measured reflections

981 independent reflections

894 reflections with  $I > 2\sigma(I)$ 

 $\mu = 0.13 \text{ mm}^{-1}$ 

T = 190 K

 $R_{\rm int} = 0.012$ 

 $\theta_{\rm max} = 27.5$ 

Crystal data

 $\begin{array}{l} C_{6}H_{12}O_{5} \\ M_{r} = 164.16 \\ \text{Orthorhombic, } P2_{1}2_{1}2_{1} \\ a = 6.3661 \ (3) \ \text{\AA} \\ b = 6.6684 \ (3) \ \text{\AA} \\ c = 17.1873 \ (9) \ \text{\AA} \\ V = 729.63 \ (6) \ \text{\AA}^{3} \end{array}$ 

### Data collection

Nonius KappaCCD area-detector diffractometer  $\omega$  scans Absorption correction: multi-scan (*DENZO/SCALEPACK*; Otwinowski & Minor, 1997)  $T_{min} = 0.88, T_{max} = 0.97$ 

## Refinement

 $\begin{array}{ll} \text{Refinement on } F^2 & w = 1/[\sigma^2(F^2) + (0.02P)^2 \\ R[F^2 > 2\sigma(F^2)] = 0.027 & + 0.17P], \\ wR(F^2) = 0.063 & \text{where } P = [\max(F_o^2, 0) + 2F_c^2]/3 \\ S = 1.01 & (\Delta/\sigma)_{\max} < 0.001 \\ 981 \text{ reflections} & \Delta\rho_{\max} = 0.17 \text{ e } \text{ Å}^{-3} \\ 100 \text{ parameters} & \Delta\rho_{\min} = -0.17 \text{ e } \text{ Å}^{-3} \\ \text{H-atom parameters constrained} \end{array}$ 

## Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$O10-H5\cdots O7^{i}$	0.84	1.95	2.750 (2)	158
O7−H9···O9 <sup>ii</sup>	0.85	2.05	2.852 (2)	158
$O9-H12 \cdot \cdot \cdot O8^{iii}$	0.84	1.86	2.694 (2)	174
$O8{-}H10{\cdot}{\cdot}{\cdot}O6^{iv}$	0.84	1.95	2.780 (2)	176

Symmetry codes: (i)  $x + \frac{1}{2}, -y + \frac{1}{2}, -z + 1$ ; (ii) x, y + 1, z; (iii)  $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$ ; (iv) x - 1, y, z.

In the absence of significant anomalous scattering, Friedel pairs were merged and the absolute configuration was assigned from the known starting materials. The H atoms were all located in a difference map, but those attached to C atoms were repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their geometry [C-H in the range 0.93–0.98 Å and O-H = 0.82 Å and  $U_{\rm iso}({\rm H})$  in the range 1.2-1.5 $U_{\rm eq}({\rm C},{\rm O})$ ], after which they were refined with riding constraints.

Data collection: *COLLECT* (Nonius, 2001); cell refinement: *DENZO/SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO/SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

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